Lake Michigan District personnel conducted approximately 75 lake monitoring surveys from 1975 to the present. The purpose of this report is to summarize the data relating nutrient concentrations, particularly dissolved or ortho-phosphorous and the inorganic forms of nitrogen  $(NH_4 + NO_3 + NO_2 \text{ as N})$ , to the productivity of a lake. How productive a lake is can be indicated by the accumulation of algae, imparting a green or brown color to the water and/or by the presence of aquatic vegetation. The more nutrients (nitrogen and phosphorous) the more abundant will be the concentration of algae or aquatic vegetation.

Nitrogen and phosphorous concentrations and the corresponding effect on algae and vegetation numbers has been documented by many researchers: (Sawyer, 1947; American Water Works Association, 1966; Volensweider, 1968; Edmondson, 1976; Lee, 1971; Ryther and Dunstan, 1971; Maloney et al., 1972; Powers et al., 1972; Martin and Goff, 1972; Shannon and Brezonik, 1972). Sawyer (1947) found phosphorous concentrations of 0.01 mg/l and nitrogen concentrations of 0.3 mg/l, particularly during spring turnover, sufficient to give the lake a high potential for an abundance of algae or aquatic vegetation.

Alkalinity or carbonate hardness, that portion of the hardness attributed to the bicarbonate plus carbonate, will be used to differentiate soft from hardwater lakes. Concentrations less than 80 mg/l as CaCO3 will be given soft water designation, 80-125 mg/l as CaCO3 moderately hard water and greater than 125 mg/l hard water. Alkalinity does not by

itself reflect the lakes productivity but indicates the geology of the watershed. A clay or limestone subsoil will traditionally create a hard water situation.

Productivity potential mentioned in the following discussion relate to Sawyer's formula. Approximately 55 lakes have been discussed, the other 20 are in those counties transferred to other districts. North Lake in Waupaca County is a 68.5 acre, 40 foot maximum depth hardwater lake. Phosphorous levels during the 1976 quarterly monitoring period remained below the Sawyer 0.01 mg/l level. Nitrogen exceeded Sawyer's levels only during the March 1, 1976 sample date; dissolved oxygen levels were also down at this time. Active decomposition of the organics under the ice would account for this. The oxygen level did remain above the 2 mg/l range, a fish kill would not be likely. Vegetation shouldn't be excessive. The bottom is marl indicating spring water seepage and little silt or muck in the littoral area.